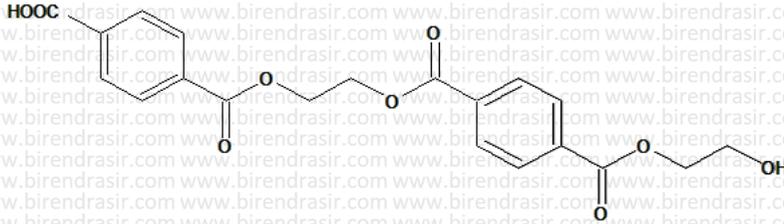


Sr. No.	NAME OF THE POLYMER	MONOMER	STRUCTURE	USES
(5)	Glyptal	(i) Ethylene Glycol (ii) Phthalic acid	$-(\text{-OCH}_2\text{-CH}_2\text{OOC} \begin{array}{c} \text{CO-} \\ \text{C}_6\text{H}_4 \end{array} \text{-})_n\text{-}$	It is used for the manufacture of paints and lacquers.
(6)	Bakelite	(i) Phenol (ii) Formaldehyde	$-(\text{-H}_2\text{C} \begin{array}{c} \text{OH} \\ \\ \text{C}_6\text{H}_3 \\ \\ \text{CH}_2 \end{array} \text{-CH}_2 \begin{array}{c} \text{OH} \\ \\ \text{C}_6\text{H}_3 \\ \\ \text{CH}_2 \end{array} \text{-})_n\text{-}$	It is used for making electrical switches, combs, computer discs, handles of utensils.
(7)	Neoprene (or Polychloroprene) Synthetic rubber	Chloroprene $\begin{array}{c} \text{Cl} \\ \\ \text{CH}_2=\text{C}-\text{CH}=\text{CH}_2 \end{array}$	$\begin{array}{c} \text{Cl} \\ \\ (-\text{CH}_2-\text{C}=\text{CH}-\text{CH}_2-) \end{array}_n$	It is used for making transmission belts, printing rolls and flexible tubing employed for conveyance of oil and petrol.

Sr. No.	NAME OF THE POLYMER	MONOMER	STRUCTURE	USES
(8)	Isoprene Natural Rubber	(2-methyl-1,3-butadiene)		-
(9)	Buna - N or GRA	(i) $\text{CH}_2=\text{CH}-\text{CH}=\text{CH}_2$ (ii) $\text{CH}_2=\text{CH}-\text{CN}$	$\left[\text{CH}_2-\text{CH}=\text{CH}-\text{CH}_2-\text{CH}_2-\underset{\text{CN}}{\text{CH}} \right]_n$ <p style="text-align: center;">BUNA-N</p>	-
(10)	Buna -S or GRS	(i) $\text{C}_6\text{H}_5\text{CH}=\text{CH}_2$ (ii) $\text{CH}_2=\text{CH}-\text{CH}=\text{CH}_2$	$\left[\text{CH}_2-\text{CH}=\text{CH}-\text{CH}_2-\underset{\text{C}_6\text{H}_5}{\text{CH}}-\text{CH}_2 \right]_n$	It is used in manufacture of tyres and other mechanical rubber goods.
(11)	Teflon	Tetrafluoroethylene ($\text{F}_2\text{C}=\text{CF}_2$)	$\left(-\text{CF}_2-\text{CF}_2- \right)_n$ <p style="text-align: center;">thermoplastic polymer</p>	Insulating material for high frequency installation.

Sr. No.	NAME OF THE POLYMER	MONOMER	STRUCTURE	USES
(12)	Nylon-66	(i) adipic acid $\text{HO}-\overset{\text{O}}{\parallel}{\text{C}}-(\text{CH}_2)_4-\overset{\text{O}}{\parallel}{\text{C}}-\text{OH}$ (ii) hexamethylene diamine $\begin{array}{c} \text{H} \\ \\ \text{H}-\text{N}-(\text{CH}_2)_6-\text{NH}_2 \end{array}$	$\left[\begin{array}{c} \text{O} \qquad \text{O H} \\ \parallel \qquad \parallel \\ \text{---C---}(\text{CH}_2)_4\text{---C---N---}(\text{CH}_2)_6\text{---NH---} \end{array} \right]$	It possess greater tensile strength, elasticity and lusture than any natural fibre. It is chemically inert and is fabricated into sheet, bristles and textile fibres.
(13)	Nylon 6 or Perolon	$\begin{array}{c} \text{NH} \\ \\ (\text{CH}_2)_5 \\ \\ \text{C=O} \end{array}$	$\left[\text{---}(\text{CH}_2)_5\overset{\text{O}}{\parallel}{\text{C}}\text{---NH---}(\text{CH}_2)_5\overset{\text{O}}{\parallel}{\text{C}}\text{---NH---} \right]$	The fibre is practically identical to Nylon in properties
(14)	Melamine polymer	(i) Melamine  (ii) HCHO formaldehyde		-

Sr. No.	NAME OF THE POLYMER	MONOMER	STRUCTURE	USES
(15)	Terylene or Dacron	(i) Ethanediol HOCH ₂ CH ₂ OH (ii) Terephthalic acid HOOC-  -COOH		-
(16)	Orlon (polyacrylonitrile)	Acrylonitrile $\text{CH}_2 = \underset{\text{CN}}{\text{CH}}$	$\text{CH}_2 - \underset{\text{CN}}{\text{CH}} - \text{CH}_2 - \underset{\text{CN}}{\text{CH}} - \text{CH}_2 - \underset{\text{CN}}{\text{CH}} - \text{CH}_2 - \underset{\text{CN}}{\text{CH}} -$	-
(17)	Polystyrene	Styrene $\text{CH}_2 = \underset{\text{C}_6\text{H}_5}{\text{CH}}$	$\text{CH}_2 - \underset{\text{C}_6\text{H}_5}{\text{CH}} - \text{CH}_2 - \underset{\text{C}_6\text{H}_5}{\text{CH}} - \text{CH}_2 - \underset{\text{C}_6\text{H}_5}{\text{CH}} -$	-
(18)	Polythene	Ethene CH ₂ =CH ₂	-(-CH ₂ -CH ₂ -) _n -	-

